Admin:

Today: Particles: interaction with flow Generation

II Particles

Heavy seeding

Number density high enough to look like a dye

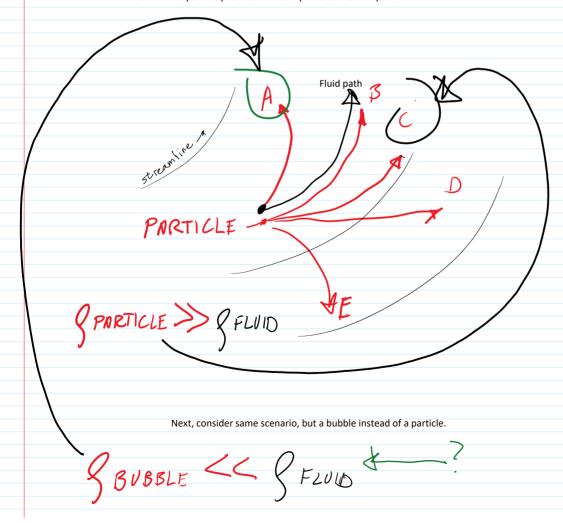
Similar considerations to dyes:

Big difference from dyes

- 1)Particles must track with the flow
 - Dyes are molecules, track with the flow just fine.
- 2)Want particles to NOT disturb flow
- 3)Want particles to show up HIGH VISIBILITY
 - 1) When will particles track well, be good tracers?

Minute paper: Consider a curved streamline in a **horizontal plane.** Consider a small particle, much denser than the fluid.

What will the particle path look like compared to the fluid path?



Ever been hit in the back of the head by a balloon when you are accelerating in a car? http://www.youtube.com/watch?v=XXpURFYgR2E

For particles (or bubbles) to track with the surrounding fluid, they must accelerate the same as the neighboring fluid

Forces on particle:

Body: gravity, neglect.

Surface: normal = pressure parallel = shear from fluid

First, assume a pressure gradient:

LOW PARTICIE HIGH PRESSURE

NET F

LOW PLUID HIGH PRESSURE

F= M OLIVER SAME NET F

Which particle will accelerate more? Newton's Second Law: $\sum F = ma$

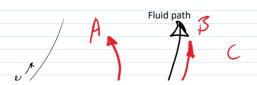
Same force - Dense - which will accelerate more?

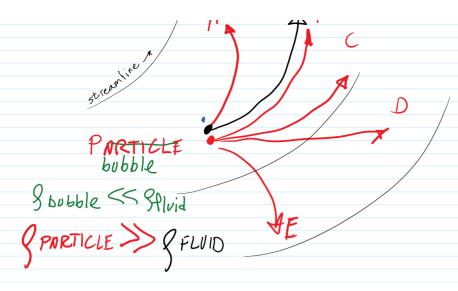
What makes streamlines curve?

(what is a streamline?)

Lon Migh

Streamlines curve because of pressure gradient. Low P is inside curve





For particles to accurately track the fluid we have

Rules of thumb:

- In water or other liquids, particles of 100
 μm diameter or less, any density, will track most flows
- In air, particles of 1 μm diameter or less, any density, will track most flows.

Similar considerations to dyes:

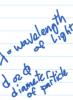
- 1) Particles must track with the flow
- 2) Want particles to NOT disturb flow
- 3)Want particles to show up HIGH VISIBILITY

2) Want particles to NOT disturb flow

- As with dyes, minimize injection differential velocity; inject at local flow speed.
- Want particles to not introduce new forces or effects. Avoid:
 - soluble particles
 - surface tension
 - o chemical reactions
 - significant change of density
 - o particle-particle interaction
 - Number density of particles = # of particles / unit volume. (Contrast to mass/volume of solid alone). Keep low enough to avoid interactions.
 - Particle-particle interaction (collisions, drag) lead to non-Newtonian effects. Slurries, oobleck, blood, shampoo, silly putty, other polymers. Gets into 'complex fluid' categories. Interesting field.

3) High visibility

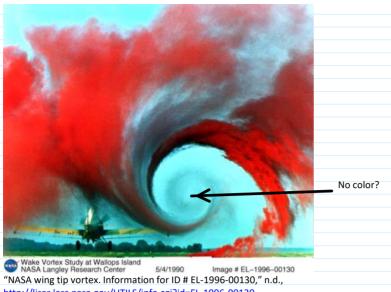
Particles only scatter light. Interaction depends on size (d) compared to λ . Scattering = \sum of reflection, refraction, diffraction & absorption



d $^{\sim}O(\lambda)$: Mie scattering regime.

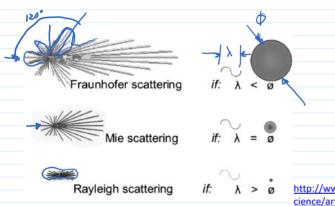
e.g. visible light =0.7 $_{-}$ 0.4 μm , so diameters of 1 μm to 0.1 μm (100 nm, 1000 A).

- o Scattering efficiency drops as particles get smaller. Better tracking, but less light.
- Independent of wavelength; no colors from particles this small. Makes clouds white.
- o Particles large enough to have color are too big to track well.



http://lisar.larc.nasa.gov/UTILS/info.cgi?id=EL-1996-00130.

Light is not scattered uniformly:

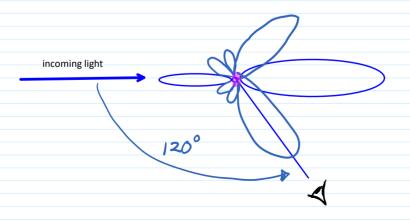


http://www.sciencedirect.com/science/article/pii/S03785173070 10113

Keck, Cornelia M., and Rainer H. Müller. "Size Analysis of Submicron Particles by Laser Diffractometry—90% of the Published Measurements Are False." International Journal of Pharmaceutics 355, no. 1–2 (May 1, 2008): 150–163. doi:10.1016/j.ijpharm.2007.12.004.



Mie regime, small particles: Back scatter < Forward scatter



Mie + Fraunhofer regime, larger particles: Back scatter < Forward scatter

Often a strong lobe at 120 degrees to incoming light. SWEET SPOT Best to play with camera-light angles.

Smaller particles, d $<< \lambda$,

Rayleigh scattering regime. Elastic collision of photons with particles. No energy exchange. Blue sky is Rayleigh scattering; sunlight scattered by molecules of air, preferentially blue. Longer wavelengths are too long to interact much; are only seen at sunset due to long passage through atmosphere, and when scattered by larger molecules of pollutants or dust.

Next: How to make or get particles

http://www.youtube.com/watch?v=DOUfyDHxkYQ&feature=related

NCFMF film 'Flow Visualization'

Hydrogen bubble technique, but also discusses streamline vs streakline vs pathline

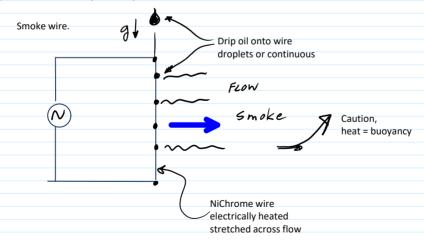
Available in lab

Aerosols in air: smoke and fog Solids liquids

Today:

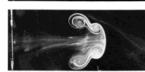
Particles - how to get/make them

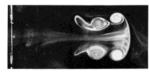
A) Smoke = soot usually, carbon particles





79. Leapfrogging of two vortex rings. Two successive part of air are ejected from an ordice of Sem damment by a patient that is driven by the inspects of two pendalmen. The flow is made visible by a smoke wire strended array or the seminar of the seminar of about 1600 based on ordice of similar the second ring graphs. At this Berwickis number of about 1600 based on ordice of similar the second ring ravels fuser in the induced field of the first, and has slipped through it in the third phenograph. Then the process is repeated, the ring slipping through the second in the list photograph. Twomber of Mannis 1876 or Mannis 1876 or photograph. Twomber of Mannis 1876 or Mannis 1876 or

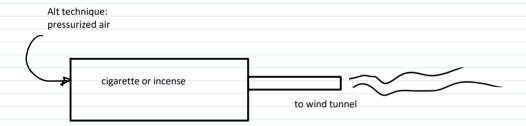






Van Dyke, Milton. *Album* of Fluid Motion. 10th ed. Parabolic Press, Inc., 1982.

Most oils work. Veg is less toxic. Generates $1\mu m$ particles. Penetrates into lungs, causes cancer, regardless of composition.



2.1. Visualization of Flow Direction and Flow Contours

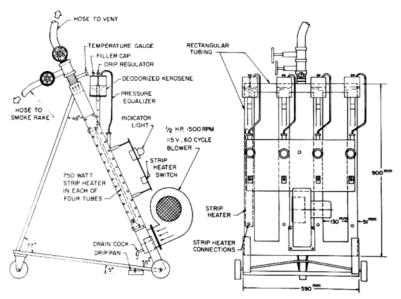


Fig. 2.6 Smoke generator designed at the University of Notre Dame. (From Mueller, 1983. Published by Hemisphere Publishing Corporation.)

Merzkirch, Wolgang. *Flow Visualization, Second Edition*. 2nd ed. Academic Press, 1987.

